
Role of Platinum Group Metals on rheological and electrical properties of nuclear glass

Muriel Neyret^{*1}, Dylan Jouglard¹, Jean Puig¹, Caroline Hanotin¹, Leire Del Campo², Mohammed Malki², and Philippe Marchal³

¹CEA Marcoule – CEA Marcoule 30207 Bagnols sur Ceze – France

²Conditions Extrêmes et Matériaux : Haute Température et Irradiation (CEMHTI) – Université d'Orléans, Centre National de la Recherche Scientifique : UPR3079 – Site Haute Température, CS 90055, 1D avenue de la Recherche Scientifique, 45071 Orléans Cedex 2, France

³Laboratoire Réactions et Génie des Procédés (LRGP) – Université de Lorraine, Centre National de la Recherche Scientifique : UMR7274 – Université de Lorraine - ENSIC, 1 rue de Grandville BP 20451, 54001 Nancy Cedex, France

Abstract

High level radioactive waste originated from nuclear fuel reprocessing are vitrified in a borosilicate matrix at 1100-1200°C. Some elements from the Platinum Group Metals (PGM) remain insoluble during the vitrification process. This might lead to complex phenomena, such as aggregation and sedimentation of particles, and affect significantly rheological and electrical properties of the glass melt. In order to master the vitrification process, a good understanding of these phenomena is necessary.

Glass electrical conductivity increases strongly when the RuO₂ content is higher than a percolation threshold, explained by a connected network of RuO₂ particles, which are metallic conductors. The formation of these aggregates is favored by particles anisotropy and by some specific process conditions (temperature, mixing,...) and leads to an increase of the glass melt viscosity.

Electrical and rheological percolation laws are found to be in good agreement, and give a good description of the PGM influence on nuclear glass properties.

Keywords: nuclear glasses, rheology, electrical conductivity, noble metals

^{*}Speaker